

行政院國家科學委員會專題研究計畫 成果報告

(Ln<sub>1-x</sub>AxMn<sub>1-y</sub>ByO<sub>3</sub>)磁傳輸特性及多層膜交換偏壓機制的  
探討

計畫類別：個別型計畫

計畫編號：NSC93-2112-M-164-002-

執行期間：93年08月01日至94年07月31日

執行單位：修平技術學院電機工程系

計畫主持人：陳宏仁

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報告類型：精簡報告

處理方式：本計畫可公開查詢

中 華 民 國 94 年 10 月 27 日

行政院國家科學委員會補助專題研究計畫  成果報告  
 期中進度報告

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執行期間：93 年 08 月 01 日至 94 年 07 月 31 日

計畫主持人：陳宏仁教授兼工程學群召集人

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執行單位：修平技術學院

中 華 民 國 94 年 10 月 26 日

# 行政院國家科學委員會專題研究計畫成果報告

## (Ln<sub>1-x</sub>A<sub>x</sub>Mn<sub>1-y</sub>ByO<sub>3</sub>)磁傳輸特性及多層膜交換偏壓機制的探討

The study of magnetotransport properties of (Ln<sub>1-x</sub>A<sub>x</sub>Mn<sub>1-y</sub>ByO<sub>3</sub>) and its exchange bias mechanism of multi-layer films

計畫編號：NSC 93-2112-M-164-002-

執行期限：93年8月1日至94年7月31日

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### 1. Abstract

此計畫主要對於 La<sub>0.7-x</sub>Ln<sub>x</sub>Pb<sub>0.3</sub>MnO<sub>3</sub> (Ln = Pr, Nd, Gd, Dy, Sm and Y) 錳化合物的磁化行為作綜合性探討。La 離子被 Pr, Nd, Gd, Dy, Sm 或 Y 取代將導致降低鐵磁有序溫度 T<sub>c</sub> 及在場冷卻(FC)-零場冷卻(ZFC)量測中顯示具一個短程有序自旋相。這主要乃相對於 La 離子 Pr, Nd, Gd, Dy, Sm 及 Y 離子具有較小的離子半徑，取代後造成鈣鈦礦結構較大的扭曲。由研究結果可獲得飽和磁化率 M<sub>s</sub> 隨著 Po 或 Na 含量的增加而增加，卻隨著 Sm 或 Y 含量的增加而降低。然而，Gd 及 Dy 含量的增加卻使得 M<sub>s</sub> 先增加後再降低。此結果可以用具 *f*-shell 電子之磁性離子及較小離子半徑兩種競爭特性來解釋鐵磁性的增加或降低。

關鍵字：鈣鈦礦結構，飽和磁化率，鐵磁性，自旋。

The magnetization behaviors of perovskite oxides, La<sub>0.7-x</sub>Ln<sub>x</sub>Pb<sub>0.3</sub>MnO<sub>3</sub> (Ln=Pr, Nd, Gd, Dy, Sm and Y), have been synthesized. The replacement of La ion by Pr, Nd, Gd, Dy, Sm or Y results in a considerable decrease in the ferromagnetic ordering temperature T<sub>c</sub> and clearly irreversible behavior in the FC-ZFC curves showing a short-range spin order phase. The fact is in agreement with the smaller ionic

radii of Pr, Nd, Gd, Dy, Sm and Y ions in contrast to La ion and the corresponding larger distortion of perovskite structures. The saturation magnetization M<sub>S</sub> increases as Pr or Nd content increases while M<sub>S</sub> decreases as Sm or Y content increases. Moreover, the saturation magnetization M<sub>S</sub> increases and then decreases as Gd and Dy content increases. These results can be explained in terms of the competition between the increase of ferromagnetically interacting spins due to the introduction of magnetic Pr or Sm ions with *f*-shell electrons and suppression of ferromagnetism due to structure tuning induced by the small ionic radius of the interpolated cation into the La-site.

Keywords : perovskite structures, saturation magnetization, ferromagnetism, spin.

### 2. Introduction

Manganese oxides, A<sub>1-x</sub>A'<sub>x</sub>MnO<sub>3</sub> (A = La, Pr, Nd, Gd, Dy, Sm, Y etc., and A' = Ca, Sr, Ba, Pb etc.), with ABO<sub>3</sub> perovskite structure have attracted considerable investigation because of the discovery of

colossal magnetoresistance (CMR) effect accompanied with rich variety of magnetic and electrical properties (1-9). The partial substitution of divalent ions for the trivalent rare earth ions induces ferromagnetism in these compounds due to the conversion of  $Mn^{3+}$  ( $t_{2g}^3 e_g^1$ ) into  $Mn^{4+}$  ( $t_{2g}^3 e_g^0$ ) explained by double exchange (DE) interaction (10-11) with additional polaronic and Jahn-Teller effects (12). These oxides show transitions from paramagnetic to ferromagnetic state in magnetism and insulating to metallic state in electricity with the decrease of temperature. There appears to be a direct relationship between the complex lattice effects and the physical properties in these perovskite compounds (13). The introduction of lanthanide ion, such as Pr, Nd, Gd, Dy, Sm and Y etc., smaller than La into the perovskite,  $La_{0.7-x}Ln_xA'_{0.3}MnO_3$ , is expected to produce a deformation of the octahedral  $MnO_6$  that results in a modified Mn-O-Mn bond angle and in a consequence reduction of the DE interaction between  $Mn^{3+}$  and  $Mn^{4+}$ . In the paper, we have made a comparative study of the influence on the magnetic properties induced by  $La^{3+}$  site substitution. To evaluate the effect of rare-earth element substitution on magnetic properties, we have studied a series of samples in which  $La^{3+}$  is partially replaced by  $Pr^{3+}$ ,  $Nd^{3+}$ ,  $Gd^{3+}$ ,  $Dy^{3+}$ ,  $Sm^{3+}$  or  $Y^{3+}$  with the  $Mn^{3+}/Mn^{4+}$  ratio fixed at 0.7/0.3 and the mean radius of the A-site ions is systematically decreased in these  $La_{0.7-x}Ln_xPb_{0.3}MnO_3$  compounds.

### 3. Experimental

Polycrystalline bulk samples of the compounds,  $La_{0.7-x}Ln_xPb_{0.3}MnO_3$  ( $Ln=Pr, Nd, Gd, Dy, Sm$  and  $Y$ ), were prepared by conventional ceramic fabrication technique of solid-state reaction. After wet-milled by SPEX 8000M, well-dried hyperfine powders,  $Ln_2O_3$  (99.95% purity,  $Ln=Nd, Gd, Dy, Sm$  and  $Y$ ),  $Pr_6O_{11}$  (99.95% purity),  $PbCO_3$  (99.99% purity) and  $MnCO_3$  (99.99% purity), were mixed by Retsch Mix Miller MM-2000 in a stoichiometric ratio and calcined in air at 800°C for 24 hours with intermediate grindings three times. After grinding, the mixed powders were pressed into a disk-shape pellet with a diameter of 12 mm and a thickness of 2 mm. The disk samples were sintered in air at 1200°C for 72 hours and then cooled down to room temperature at a cooling rate of 3°C/min. The structure and phase purity of the samples were examined by X-ray diffraction with Cu-K  $\alpha$  radiation. The magnetization measurements between 5 K and 350 K were performed in a quantum designed superconducting quantum interference device MPMS-5S SQUID magnetometer. The zero-field cooling (ZFC) and field cooling (FC) curves were taken under an applied field of 100 Oe. Temperature dependence of magnetization curves in an applied field of 50 kOe were also recorded from 5 K to 350 K.

### 4. Results and Discussion

The magnetization behaviors of perovskite oxides,  $La_{0.7-x}Ln_xPb_{0.3}MnO_3$  ( $Ln=Pr, Nd, Gd, Dy, Sm$  and  $Y$ ), have been

synthesized. The size mismatch effect occurs when the Ln-site ions are smaller than La to fill the space in the octahedral  $\text{MnO}_6$  and, consequently, results in a more distorted structure and Mn-O-Mn bond bending (14). The ionic radii and electronic configuration of Pr, Nd, Gd, Dy, Sm and Y are listed in Table 1 (15). Hence, it is interest for us to survy the distortion effect induced by La-site substitution on the magnetic behavior.

In order to exam the spin order and magnetic behavior, we measured the ZFC and FC magnetization curves in a low field of 100 Oe. Low field ZFC-FC magnetization measurement is one of the methods generally used to characterize short-range spin order or long-range spin order behaviors. The ZFC and FC curves of all samples are presented in Fig. 1. For  $x = 0$  compounds of all systems, the ZFC and FC magnetization curves, except at very low temperature, are almost overlap indicating a ferromagnetic long-range spin order. Conversely, the irreversibility is found for Ln-site substituted compounds, and indicates the existence of short-range spin order. In these Ln-site substituted compounds, the ZFC curves coincide with the FC curves at high temperature, but separate as ths temperature decreases below ferromagnetic-paramagnetic transition temperature  $T_C$  (defined as the temperature where the slop,  $|dM/dT|$ , reaches a maximum value calculated from the ZFC curve). In pure manganite, the local clustering of the  $\text{Mn}^{3+}$  ions around the  $\text{Mn}^{4+}$  ones eventually leads to a ferromagnetic spin order of long-range type. For the Ln-site substituted phases, the smaller radius of the

introduced Ln cation generates a distortion in the cell which leads the zigzagging of the  $\text{MnO}_6$  octahedra containing the manganese atoms. As a result, the Mn-O-Mn angle decreases and the ferromagnetic DE interaction suppresses. Thus, the substitution of La by Pr, Nd, Gd, Dy, Sm or Y results in a considerable decrease of  $T_C$  due to the effect of lattice distortion and Mn-O-Mn bonds bending induced by substituting smaller lanthanide ions for La ion.

Figure 2 shows the temperature dependence of magnetization curves measured in an applied field of 50 kOe for the  $\text{La}_{0.7-x}\text{Nd}_x\text{Pb}_{0.3}\text{MnO}_3$  and  $\text{La}_{0.7-x}\text{Pr}_x\text{Pb}_{0.3}\text{MnO}_3$  oxide systems. The samples undergo a paramagnetic to ferromagnetic transition as the temperature decreases. For both of the two systems, the saturation magnetization ( $M_S$ , defined as the magnetization at 5 K and an applied field of 50 kOe for comparison) increases with the increase of Nd or Pr content. In the Nd-doped compounds, the manganese sublattice is ferromagnetic and the magnetic ions,  $\text{Nd}^{3+}$ , with *f*-shell electrons are gradually aligned with manganese as the temperature decreases. The rare earth magnetic ions  $\text{Nd}^{3+}$  contribute to an additional magnetization value to the total moment for Nd-doped samples. However, the  $\text{Nd}^{3+}$ , with *f*-shell electrons (electronic configuration as listed in Table 1) are hard to be fully aligned with the manganese due to the deformation induced by the substitution of Nd for La (radii as listed in Table 1). As Fig. 2(a) shows, complete magnetic saturation for Nd-doped samples is not achieved even at 5 K, which

shows high  $dM/dT$  slope. For the Pr-doped compounds, similar behavior was also observed. However, due to the less unpaired  $f$ -shell electrons of  $\text{Pr}^{3+}$  than  $\text{Nd}^{3+}$ , the  $M_S$  values are smaller of Pr-doped compounds than those of Nd-doped ones as shown in Fig. 2(a) and (b).

The temperature dependence of magnetization curves measured in an applied field of 50 kOe for the  $\text{La}_{0.7-x}\text{Gd}_x\text{Pb}_{0.3}\text{MnO}_3$  and  $\text{La}_{0.7-x}\text{Dy}_x\text{Pb}_{0.3}\text{MnO}_3$  oxide systems are shown in Fig. 3. The samples also perform a paramagnetic to ferromagnetic transition as the temperature decreases. For both of the two systems, the  $M_S$  increases initially with the increase of Gd or Dy content, reaches a maximum value for  $x = 0.1$  and then decreases with the increase of Gd or Dy content. The magnetic behaviors of Gd-doped and Dy-doped are dissimilar from that of Nd-doped and Pr-doped. The results can be explained in terms of the competition between the increase of ferromagnetically interacting spins and suppression of ferromagnetism. As listed in Table 1, the effective moment values of  $\text{Gd}^{3+}$  and  $\text{Dy}^{3+}$  are quite larger than that of  $\text{La}^{3+}$ ,  $\text{Pr}^{3+}$  and  $\text{Nd}^{3+}$  and in a consequence the  $\text{Gd}^{3+}$  or  $\text{Dy}^{3+}$  contribute to a more additional magnetization value to the total moment than  $\text{Pr}^{3+}$  and  $\text{Nd}^{3+}$ . This is the reason why  $M_S$  increases with the increase of Gd or Dy content due to the increase of ferromagnetically interacting spins initially. However, the radii of  $\text{Gd}^{3+}$  and  $\text{Dy}^{3+}$  are quite smaller than  $\text{La}^{3+}$ ,  $\text{Pr}^{3+}$  and  $\text{Nd}^{3+}$ . The introduction of Gd or Dy into La site induces heavy distortion of crystal structure and suppression of ferromagnetism.

Complete magnetic saturation for Gd-doped and Dy-doped samples are not achieved even at 5 K. The higher  $dM/dT$  slope values for heavy doped samples show that the suppression of ferromagnetism induced by structure distortion plays a primary role. Thus, the  $M_S$  decreases with the increase of doped content for  $x \geq 0.1$ .

In contrast,  $M_S$  decreases with the increase of Sm or Y content in the Sm-doped and Y-doped oxides as shown in Fig. 4. It can be inferred that the substitution of smallest  $\text{Sm}^{3+}$  or  $\text{Y}^{3+}$  for  $\text{La}^{3+}$  can distort the perovskite structure and give rise to a larger bending angle of the Mn-O-Mn, which consequently weakens the ferromagnetic double exchange interaction between  $\text{Mn}^{3+}$  and  $\text{Mn}^{4+}$ . The different magnetization processes of the two system can be ascribed to a competition in these compounds between suppression of ferromagnetism induced by lattice distortion and increase of ferromagnetically interacting spins due to the introduction of magnetic  $\text{Ln}^{3+}$  ions with  $f$ -shell electrons. The effective magnetic moments are  $0.84 \mu_B$  for  $\text{Sm}^{3+}$  and  $0 \mu_B$  for  $\text{Y}^{3+}$ , which is much smaller than those of  $\text{Pr}^{3+}$ ,  $\text{Nd}^{3+}$ ,  $\text{Gd}^{3+}$  and  $\text{Dy}^{3+}$ . Therefore, the increase of ferromagnetism suppression induced by structure distortion with the increase of Sm or Y content plays a major role in the competition for Sm-doped and Y-doped compounds and which is quite different from that of the rest compounds.

## 5. Conclusions

In summary, we have carried out the

magnetization measurement of the perovskite oxides,  $\text{La}_{0.7-x}\text{Ln}_x\text{Pb}_{0.3}\text{MnO}_3$  (Ln=Pr, Nd, Gd, Dy, Sm and Y) to study the substitution effect of La-site on the magnetic behavior. The magnetization behavior of (Nd, Pr)-doped systems is dissimilar to (Gd, Dy)-doped or (Sm, Y)-doped systems. The dissimilar magnetic behavior of these three systems can be interpreted in terms of the competition between increase of ferromagnetically interacting spins due to the introduction of magnetic lanthanide ions with *f*-shell electrons and suppression of ferromagnetism induced by lattice distortion. This work was sponsored by the National Science Council of the Republic of China under the grants No. NSC 93-2112-M-164-002-.

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Table 1 Values of radius, electronic configuration, effective moment and magnetism for trivalent lanthanide and yttrium ions (15).

Ion	Radius (nm)	Electronic configuration	Effective moment ( $\mu_B$ )	Magnetism
La <sup>3+</sup>	0.136	[Xe]	0	nonmagnetic
Pr <sup>3+</sup>	0.130	[Xe](4f) <sup>2</sup>	3.58	magnetic
Nd <sup>3+</sup>	0.127	[Xe](4f) <sup>3</sup>	3.62	magnetic
Sm <sup>3+</sup>	0.124	[Xe](4f) <sup>5</sup>	0.84	magnetic
Gd <sup>3+</sup>	0.122	[Xe](4f) <sup>7</sup>	7.94	magnetic
Dy <sup>3+</sup>	0.120	[Xe](4f) <sup>9</sup>	10.63	magnetic
Y <sup>3+</sup>	0.119	[Kr]	0	nonmagnetic

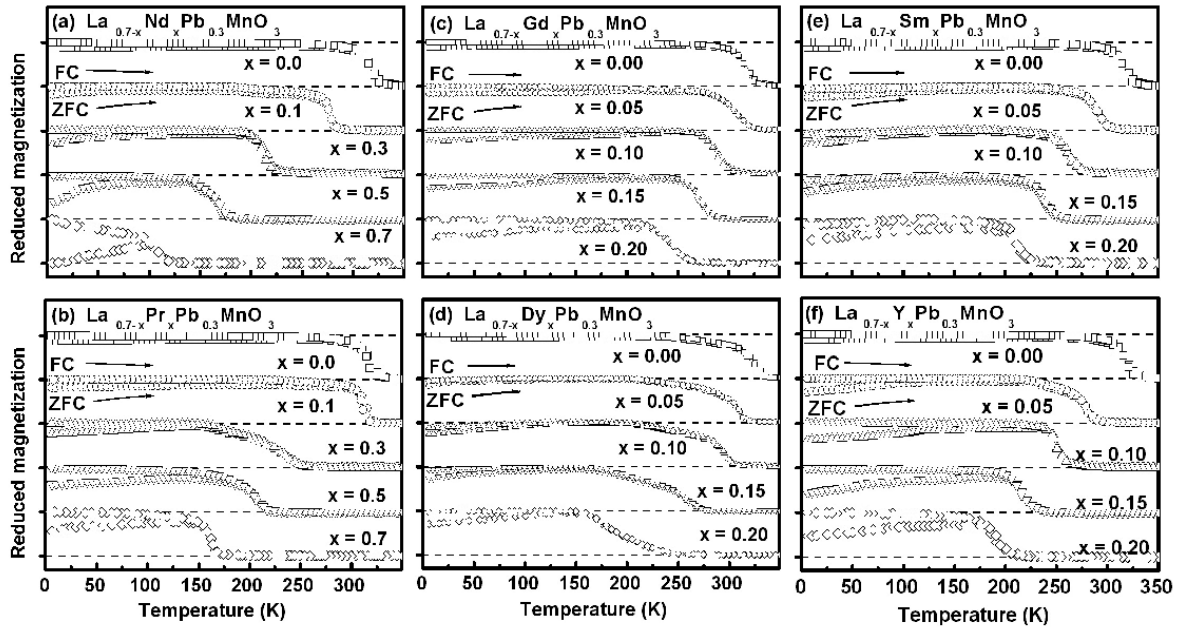


Fig. 1 Temperature dependence of zero-field cooling and field cooling magnetization curves for all systems.



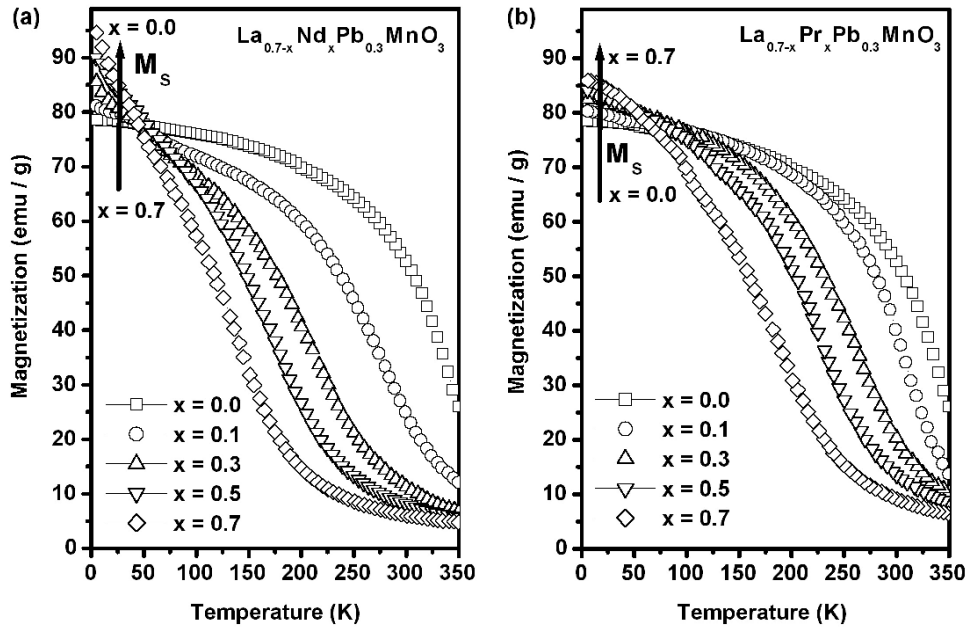


Fig. 2 Temperature dependence of magnetization curves measured in an applied field of 50 KOe for the (a) $\text{La}_{0.7-x}\text{Nd}_x\text{Pb}_{0.3}\text{MnO}_3$  and (b) $\text{La}_{0.7-x}\text{Pr}_x\text{Pb}_{0.3}\text{MnO}_3$  oxide systems.

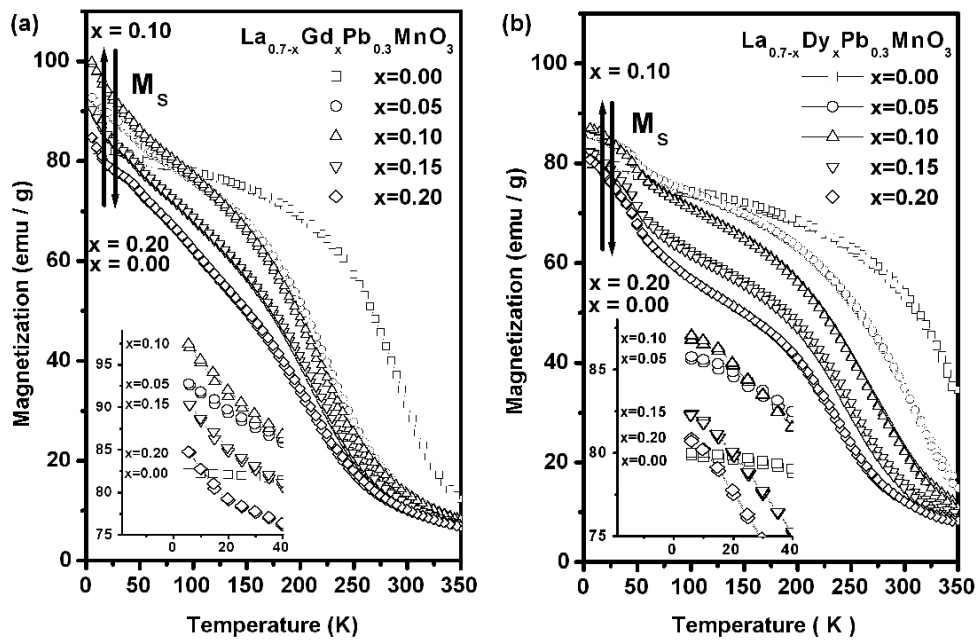


Fig. 3 Temperature dependence of magnetization curves measured in an applied field of 50 KOe for the (a) $\text{La}_{0.7-x}\text{Gd}_x\text{Pb}_{0.3}\text{MnO}_3$  and (b) $\text{La}_{0.7-x}\text{Dy}_x\text{Pb}_{0.3}\text{MnO}_3$  oxide systems.

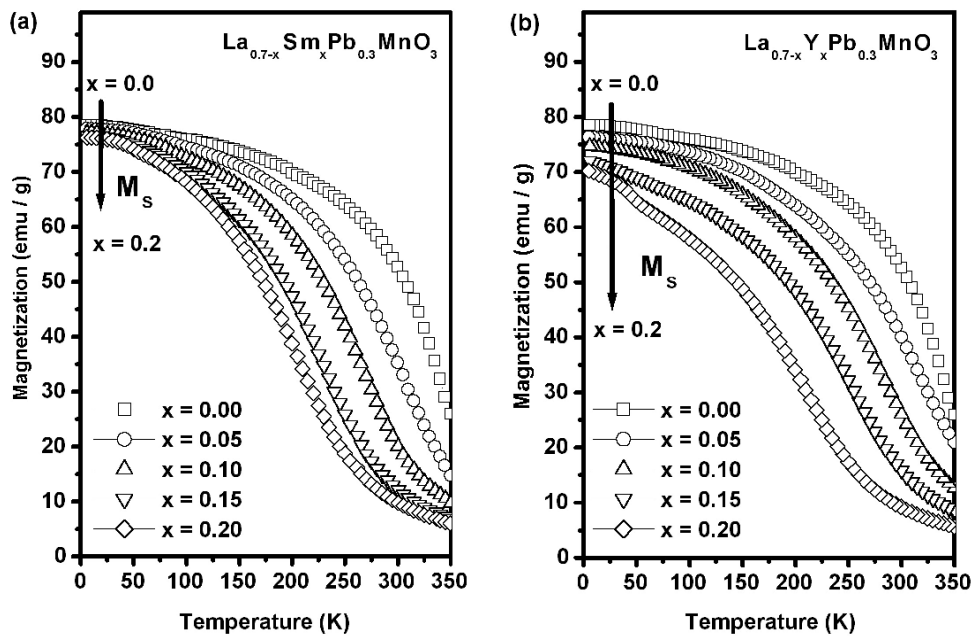


Fig. 4 Temperature dependence of magnetization curves measured in an applied field of 50 KOe for the (a) $\text{La}_{0.7-x}\text{Sm}_x\text{Pb}_{0.3}\text{MnO}_3$  and (b) $\text{La}_{0.7-x}\text{Y}_x\text{Pb}_{0.3}\text{MnO}_3$  oxide systems.